# ANNUAL WATER OUALITY REPORT

WATER TESTING PERFORMED IN 2015

Presented By Southington Water Department

### Meeting the Challenge

Once again we are proud to present our annual drinking water report, covering all drinking water testing performed between January 1 and December 31, 2015. Over the years, we have dedicated ourselves to producing drinking water that meets all state and federal standards. We continually strive to adopt new methods for delivering the best quality drinking water to your homes and businesses. As new challenges to drinking water safety emerge, we remain vigilant in meeting the goals of source water protection, water conservation, and community education while continuing to serve the needs of all of our water users.

Please remember that we are always available to assist you, should you ever have any questions or concerns about your water.

### Important Health Information

Sources of lead in drinking water includes corrosion of household plumbing system and erosion of natural deposits. Infants and children who drink water containing lead in excess of the action level could experience delays in their physical or mental development. Children could show slight deficits in attention span and learning abilities. Adults who drink this water over many years could develop kidney problems or high blood pressure.

Sources of copper in drinking water includes corrosion of household plumbing system, erosion of natural deposits and leaching from wood preservatives. Copper is an essential nutrient, but some people who drink water containing copper in excess of the action level over a relatively short amount of time could experience gastrointestinal distress. Some people who drink water containing copper in excess of the action level over many years could suffer liver or kidney damage. People with Wilson's Disease should consult their personal doctor.

Some people may be more vulnerable to contaminants in drinking water than the general population. Immunocompromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants may be particularly at risk from infections. These people should seek advice about drinking water from

their health care providers. The U.S. EPA/CDC (Centers for Disease Control and Prevention) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline at (800) 426-4791.



### Substances That Could Be in Water

To ensure that tap water is safe to drink, the U.S. EPA prescribes regulations limiting the amount of certain contaminants in water provided by public water systems. U.S. Food and Drug Administration regulations establish limits for contaminants in bottled water, which must provide the same protection for public health. Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of these contaminants does not necessarily indicate that the water poses a health risk.

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals, in some cases, radioactive material, and substances resulting from the presence of animals or from human activity. Substances that may be present in source water include:

**Microbial Contaminants**, such as viruses and bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations, or wildlife;

**Inorganic Contaminants**, such as salts and metals, which can be naturally occurring or may result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming;

**Pesticides and Herbicides**, which may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses;

**Organic Chemical Contaminants**, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production and may also come from gas stations, urban stormwater runoff, and septic systems;

**Radioactive Contaminants**, which can be naturally occurring or may be the result of oil and gas production and mining activities.

For more information about contaminants and potential health effects, call the U.S. EPA's Safe Drinking Water Hotline at (800) 426-4791.

# How Long Can I Store Drinking Water?

The disinfectant in drinking water will eventually dissipate even in a closed container. If that container housed bacteria prior to filling up with the tap water the bacteria may continue to grow once the disinfectant has dissipated. Some experts believe that water could be stored up to six months before needing to be replaced. Refrigeration will help slow the bacterial growth.

### Water Conservation

You can play a role in conserving water and saving yourself money in the process by becoming conscious of the amount of water your household is using and by looking for ways to use less whenever you can. It is not hard to conserve water. Here are a few tips:

- Automatic dishwashers use 15 gallons for every cycle, regardless of how many dishes are loaded. So get a run for your money and load it to capacity.
- Turn off the tap when brushing your teeth.
- Check every faucet in your home for leaks. Just a slow drip can waste 15 to 20 gallons a day. Fix it and you can save almost 6,000 gallons per year.
- · Check your toilets for leaks by putting a few drops of food coloring in the tank. Watch for a few minutes to see if the color shows up in the bowl. It is not uncommon to lose up to 100 gallons a day from an invisible toilet leak. Fix it and you save more than 30,000 gallons a year.
- Use your water meter to detect hidden leaks. Simply turn off all taps and water using appliances. Then check the meter after 15 minutes. If it moved, you have a leak.

### Where Does My Water Come From?

The Southington Water Works Department L supplies its customers with a mixture of surface water and ground water. These water sources include three reservoirs (Southington Reservoir #1,

Southington Reservoir #2, and Southington Reservoir #3) on the Southington-Wolcott town line, and six groundwater wells located throughout Southington. Each of these sources is treated specifically based on the needs of the water before becoming available to the public through a vast network of underground pipelines that lie below the town.



### Source Water Protection

The Southington Water Department owns and controls more than 90% of its watershed land and follows best management practices for source water protection. This includes annual watershed inspection and an active forestry management program. The Southington Water Department has also implemented frequent patrols of watershed land and aquifer protection areas to reduce trespassing.



### **Community Participation**

We encourage public interest and participation in our community decisions affecting water. Regular meetings of the Southington Board of Water Commissioners occur once each month. The public is welcome to attend these meetings. A complete listing of meeting locations, dates, and times can be obtained by calling the Town Clerk Office at (860) 276-6211, or our office at (860) 628-5593.

### Information on the Internet

The U.S. EPA (www.epa.gov/Your-Drinking-Water) and the Centers for Disease Control and Prevention (www.cdc.gov) Web sites provide a substantial amount of information on many issues relating to water resources, water conservation and public health. Also, the Connecticut Department of Public Health has a Web site (www.ct.gov/dph/ cwp/view.asp?a=3139&q=535164) that provides complete and current information on water issues in Connecticut, including valuable information about our watershed.

### **OUESTIONS?**

For more information about this report, or for any questions relating to your drinking water, please call Frederick W. Rogers, Superintendent, at (860) 628-5593.

### Lead in Home Plumbing

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. We are responsible for providing high-quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at www.epa.gov/lead.

### Is tap water cheaper than soda?

Yes! You can refill an 8 oz. glass of tap water approximately 15,000 times for the same cost as a six-pack of soda pop. And, water has no sugar or caffeine.

### How long can a person go without water?

Although a person can live without food for more than a month, a person can only live without water for approximately one week.

### When was drinking water first regulated?

The Safe Drinking Water Act (SDWA) of 1974 represents the first time that public drinking water supplies were protected on a federal (national) level in the U.S. Amendments were made to the SDWA in 1986 and 1996.

## Seventy-one percent of Earth is covered in water: how much is drinkable?

Oceans hold about 96.5 percent of all Earth's water. Only three percent of the earth's water can be used as drinking water. Seventy-five percent of the world's fresh water is frozen in the polar ice caps.

#### How much water do we use every day?

The average person in the U.S. uses 80 to 100 gallons of water each day. (During medieval times a person used only 5 gallons per day.) It takes 2 gallons to brush your teeth, 2 to 7 gallons to flush a toilet, and 25 to 50 gallons to take a shower.

### When was chlorine first used in the U.S.?

In 1908, Jersey City, New Jersey and Chicago, Illinois were the first water supplies to be chlorinated in the U.S.

### Sampling Results

During the past year, we have taken hundreds of water samples to determine the presence of any radioactive, biological, inorganic, volatile organic, or synthetic organic organic organic, the table below shows only those contaminants that were detected in the water. The state requires us to monitor for certain substances less than once per year because the concentrations of these substances do not change frequently. In these cases, the most recent sample data are included, along with the year in which the sample was taken.

We participated in the 3rd stage of the EPA's Unregulated Contaminant Monitoring Rule (UCMR3) program by performing additional tests on our drinking water. UCMR3 benefits the environment and public health by providing the EPA with data on the occurrence of contaminants suspected to be in drinking water, in order to determine if EPA needs to introduce new regulatory standards to improve drinking water quality. Contact us for more information on this program

| REGULATED SUBSTANCES   |                        |                 |       |               |  |                        |                    |                    |                               |                   |                |   |                      |  |
|--|------------------------|-----------------|-------|---------------|--|------------------------|--------------------|--------------------|-------------------------------|-------------------|----------------|---|----------------------|--|
| SUBSTANCE<br>(UNIT OF MEASURE)   |                        | YEAR<br>SAMPLED |       | MCL<br>[MRDL] | MCLG<br>[MRDLG]                          | AMOUNT<br>DETECTED     |                    | RANGE<br>LOW-HIGH  |                               | DLATION           | TYPICAL SOURCE |   |                      |  |
| Alpha Emitters (pCi/L)   |                        | 201             | 2     | 15            | 0  | 2.85                   | 2.85 0.63–2.85     |                    |                               | No                | Erosion o      | Erosion of natural deposits   |                      |  |
| Barium (ppm)   |                        |                 | 2015  |               | 2  | 2                      | 0.35               | C                  | 0.0084–0.35                   |                   | No             | Discharge of drilling wastes; Discharge from metal refineries; Erosion of natural deposits                                |                      |  |
| <b>Beta/Photon Emitters</b> <sup>1</sup> (pCi/L)   |                        |                 | 201   | 2             | 50                                       | 0                      | 3.18               |                    | 0.65-3.18                     |                   | No             | Decay of  | natural and man-n    | nade deposits                              |
| Chlorine (ppm)   |                        |                 | 201   | 5             | [4]                                      | [4]                    | 0.99               |                    | 0.35–0.99                     |                   | No             | Water ad  | ditive used to contr | ol microbes                                |
| Chromium (ppb)   |                        |                 | 201   | 5             | 100                                      | 100                    | 0.0019             | )                  | ND-1.9                        |                   | No             | Discharge from steel and pulp mills; Erosion of natural deposits  |                      |  |
| Combined Radium (pCi/L)  |                        |                 | 201   | 2             | 5  | 0                      | 1.07               |                    | 0.3-1.07                      |                   | No             | Erosion of  | of natural deposits  |  |
| Fluoride (ppm)   |                        |                 | 2015  |               | 4  | 4                      | 1.55               | 0.35–1.55          |                               |                   | No             | Erosion of natural deposits; Water additive which promotes strong teeth; Discharge from fertilizer and aluminum factories |                      |  |
| Haloacetic Acids [HAA] (ppb)   |                        |                 | 2015  |               | 60                                       | NA                     | 10.79              |                    | 5.94-10.79                    |                   | No             | By-product of drinking water disinfection   |                      |  |
| <b>Nitrate</b> <sup>3</sup> (ppm)  |                        |                 | 2015  |               | 10                                       | 10                     | 5.1                |                    | 2.8–5.1                       |                   | No             | Runoff from fertilizer use; Leaching from septic tanks, sewage; Erosion of natural deposits                               |                      |  |
| TTHMs [Total Trihalomethanes] (ppb)  |                        |                 | 2015  |               | 80                                       | NA                     | 22.61              | 1                  | 17.18–22.61                   |                   | No             | By-product of drinking water disinfection   |                      |  |
| Turbidity <sup>2</sup> (NTU)   |                        |                 | 201   | 5             | ΤT                                       | NA                     | 0.128              | 0.128 0.071–0.12   |                               |                   | No             | Soil runoff   |                      |  |
| <b>Turbidity</b> (Lowest monthly percent of samples meeting limit)                                       |                        | cent of         | samj  |               | TT = 95% of<br>samples < or<br>= 0.3 NTU | NA                     | 100                | 100 NA             |                               |                   | No             | Soil runoff   |                      |  |
| Tap water samples were collected for lead and copper analyses from sample sites throughout the community |                        |                 |       |               |  |                        |                    |                    |                               |                   |                |   |                      |  |
| SUBSTANCE YEAR<br>(UNIT OF MEASURE) SAMPLED  |                        |                 |       |               |  | IT DETECTE<br>TH%TILE) |                    |                    | BOVE AL/<br>L SITES VIOLATION |                   | TYPICAL SOURCE |   |                      |  |
| Copper (ppm)   | 201                    | 14              | 1.3   | 1.            | 3  | ND                     | 0/                 |                    | 0                             | No                |                | Corrosion of household plumbing systems; Erosion of natural deposits  |                      |  |
| Lead (ppb)   | <b>Lead</b> (ppb) 2014 |                 | 15    | 0             |  | 0.42                   | 1/30               |                    | 0                             | Ν                 | No Corrosio    |   | n of household plur  | nbing systems; Erosion of natural deposits |
| OTHER REGULATED SUBSTANCES   |                        |                 |       |               |  |                        |                    |                    |                               |                   |                |   |                      |  |
|  |                        | AR<br>PLED      |       | MCL<br>[MRDL] |  | MCLG<br>[MRDLG]        |                    | AMOUNT<br>DETECTED |                               | RANGE<br>LOW-HIGH |                | VIOLATION   | TYPICAL SOURCE       |  |
| Strontium 90 (pCi/L)   | trontium 90 (pCi/L) 20 |                 | 012   |               | 8  | NA                     | A                  |                    | 1.32                          |                   | 1.32-1.32      |   | No                   | Nuclear fission                            |
| Tritium (pCi/L)  | Tritium (pCi/L)20      |                 | 12 20 |               | 20,000                                   | 0 NA                   |                    | 607.2              |                               |                   | 128.3-607.2    |   | No                   | NA   |
| SECONDARY SUBSTANCES   |                        |                 |       |               |  |                        |                    |                    |                               |                   |                |   |                      |  |
| SUBSTANCE<br>(UNIT OF MEASURE)   | YEAR<br>ASURE) SAMPLED |                 | SMCL  |               | MCLG                                     |                        | AMOUNT<br>DETECTED |                    | RANGE<br>LOW-HIGH             |                   | VIOLATI        |   | TYPICAL SOURCE       |  |
| Chloride (ppm)   | 2015                   |                 | 250   |               | NA                                       |                        | 120                |                    | 13–120                        |                   | No             |   | Runoff/leaching fr   | om natural deposits                        |
| Sulfate (ppm)  | ppm) 2015              |                 | 250   |               | NA                                       |                        | 38                 |                    | ND-38                         |                   | No             |   | Runoff/leaching fr   | om natural deposits; Industrial wastes     |

| UNREGULATED SUBSTANCES            |                 |                    |                   |   |  |  |  |  |
|-----------------------------------|-----------------|--------------------|-------------------|---|--|--|--|--|
| SUBSTANCE<br>(UNIT OF MEASURE)    | YEAR<br>SAMPLED | AMOUNT<br>DETECTED | RANGE<br>LOW-HIGH | TYPICAL SOURCE                            |  |  |  |  |
| Bromodichloromethane (ppb)        | 2015            | 13                 | 0.62–13           | By-product of drinking water disinfection |  |  |  |  |
| Bromoform (ppb)                   | 2015            | 3.3                | ND-3.3            | By-product of drinking water disinfection |  |  |  |  |
| Chloroform (ppb)                  | 2015            | 84                 | ND-84             | By-product of drinking water disinfection |  |  |  |  |
| <b>Dibromochloromethane</b> (ppb) | 2015            | 4.8                | 1.0-4.8           | By-product of drinking water disinfection |  |  |  |  |
| Metolachlor (ppb)                 | 2015            | 0.52               | ND-0.52           | NA  |  |  |  |  |
| Sodium (ppm)                      | 2015            | 46                 | 8.6–46            | Naturally occurring; Road salt            |  |  |  |  |

 $^1 {\rm The}$  MCL for beta particles is 4 mrem/year. U.S. EPA considers 50 pCi/L to be the level of concern for beta particles.

<sup>2</sup>Turbidity is a measure of the cloudiness of the water. It is monitored because it is a good indicator of the effectiveness of the filtration system.

<sup>3</sup>Nitrate in drinking water at levels above 10 ppm is a health risk for infants of less than six months of age. High nitrate levels in drinking water can cause blue baby syndrome. Nitrate levels may rise quickly for short periods of time because of rainfall or agricultural activity. If you are caring for an infant you should ask advice from your health care provider.

### **Definitions**

AL (Action Level): The concentration of a contaminant which, if exceeded, triggers treatment or other requirements which a water system must follow.

LRAA (Locational Running Annual Average): The average of sample analytical results for samples taken at a particular monitoring location during the previous four calendar quarters. Amount Detected values for TTHMs and HAAs are reported as LRAAs.

MCL (Maximum Contaminant Level): The highest level of a contaminant that is allowed in drinking water. MCLs are set as close to the MCLGs as feasible using the best available treatment technology.

MCLG (Maximum Contaminant Level Goal): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs allow for a margin of safety.

MRDL (Maximum Residual Disinfectant Level): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

MRDLG (Maximum Residual Disinfectant Level Goal): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

NA: Not applicable

ND (Not detected): Indicates that the substance was not found by laboratory analysis.

NTU (Nephelometric Turbidity Units): Measurement of the clarity, or turbidity, of water. Turbidity in excess of 5 NTU is just noticeable to the average person.

pCi/L (picocuries per liter): A measure of radioactivity.

ppb (parts per billion): One part substance per billion parts water (or micrograms per liter).

ppm (parts per million): One part substance per million parts water (or milligrams per liter).

SMCL (Secondary Maximum Contaminant Level): SMCLs are established to regulate the aesthetics of drinking water like appearance, taste and odor.

TT (Treatment Technique): A required process intended to reduce the level of a contaminant in drinking water.